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Léon Dupont, a* Smail Khelili, b Bernard Pirotte, b Bernard Tinant and Pascal De Tullio b

^aUnité de Cristallographie, Institut de Physique -Bât. B5, Université de Liège, Allée du 6 août, 17, B-4000 Liège, Belgium, ^bService de Chimie Pharmaceutique, Institut de Pharmacie - Bât. B36, Université de Liège, Avenue de l'Hôpital, 1, B-4000 Liège, Belgium, and ^cLaboratoire de Chimie Physique et de Cristallographie, Unité CPMC, Université Catholique de Louvain, Place Louis Pasteur, 1, B-1348 Louvain-la-Neuve, Belgium

Correspondence e-mail: leon.dupont@ulg.ac.be

Key indicators

Single-crystal X-ray study $T=293~\mathrm{K}$ Mean $\sigma(\mathrm{C-C})=0.004~\mathrm{\mathring{A}}$ R factor = 0.053 wR factor = 0.149 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Absolute configuration of N-[(R)-3-methyl-2-butyl]-(S)-2-(6-methoxy-2-naphthyl)propionamide

The title compound, $C_{19}H_{25}NO_2$, was made in order to confirm the absolute configuration of (R)-3-methyl-2-butylamine obtained from commercially available (R/S)-3-methyl-2-butylamine (racemate) after fractional crystallization of its ammonium salt with the optically pure carboxylic acid derivative, (S)-naproxene.

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Comment

(R)-3-Methyl-2-butylamine reacted with succinimidyl (S)-(6-methoxy-2-naphthyl)propionate to form the corresponding amide, *i.e.* the title compound, (I). The Flack (1983) parameter -0.3 (12) calculated after the crystal structure refinement does not allow us to determine reliably the absolute configuration. Nevertheless, as the product was prepared from (S)-naproxene, the C12A and C12B atoms have an S configuration and the chirality of C15A and C15B, deduced from the torsional angles, is R. The stacking of the molecules is the result of van der Waals contacts, and of N1A-H···O2B and N1B-H···O2Aⁱ [symmetry code: (i) 1 + x, y, 1 + z] hydrogen bonds. Molecules A and B form infinite ABABAB... chains in a direction parallel to the diagonal of the a and c axes.

$$H_3C$$
(I)

Experimental

The preparation of the title compound is described by Khelili *et al.* (1999). Crystals were obtained by slow evaporation of a methanol solution.

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Crystal data

 $C_{19}H_{25}NO_2$ $M_r = 299.40$ Monoclinic, P_{21} a = 6.879 (2) Å b = 29.563 (9) Å c = 8.511 (3) Å $\beta = 99.11$ (2)° V = 1709.0 (9) Å³ Z = 4 D_x = 1.164 Mg m⁻³ Mo $K\alpha$ radiation Cell parameters from 133 reflections θ = 2.5–24.5° μ = 0.08 mm⁻¹ T = 293 (2) K Prism, colourless 0.34 × 0.23 × 0.23 mm

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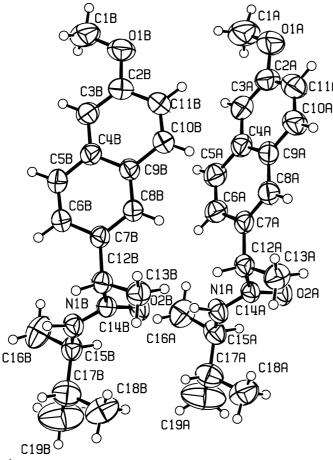


Figure 1

The molecular structure with atom-labelling schemes of molecules A and B. Displacement ellipsoids are shown at 50% probability levels and H atoms are drawn as small circles of arbitrary radii.

 $R_{\rm int} = 0.056$

 $\theta_{\rm max} = 24.4^{\circ}$

 $h = -7 \rightarrow 7$

 $k = -34 \rightarrow 34$

 $= -9 \rightarrow 9$

Data collection

MAR345 image-plate diffractometer 222 images, 150 mm, $\Delta\Phi$ 2° scans 19 313 measured reflections 5448 independent reflections 4896 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.053$ $wR(F^2) = 0.150$ S = 1.065448 reflections 414 parameters H atoms treated by a mixture of independent and constrained refinement
$$\begin{split} w &= 1/[\sigma^2(F_o^2) + (0.1033P)^2 \\ &+ 0.0697P] \\ \text{where } P &= (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{\text{max}} &< 0.001 \\ \Delta\rho_{\text{max}} &= 0.23 \text{ e Å}^{-3} \\ \Delta\rho_{\text{min}} &= -0.20 \text{ e Å}^{-3} \\ \text{Extinction correction: } SHELXL97 \\ \text{Extinction toefficient: } 0.171 (12) \\ \text{Absolute structure: Flack (1983);} \\ 1311 \text{ Friedel pairs} \\ \text{Flack parameter } &= -0.3 (12) \end{split}$$

 Table 1

 Selected geometric parameters (\mathring{A} , °).

C1A – O1A	1.414 (4)	C1 <i>B</i> —O1 <i>B</i>	1.418 (3)
C2A - O1A	1.380 (3)	C2B-O1B	1.369 (3)
C14A - O2A	1.228 (3)	C14B-O2B	1.229 (3)
C14A – N1A	1.332 (3)	C14B-N1B	1.337 (3)
C15A – N1A	1.475 (3)	C15B-N1B	1.463 (3)
O2A - C14A - N1A	124.4(2)	O2B - C14B - N1B	123.3 (2)
O2A - C14A - C12A	121.2 (2)	O2B - C14B - C12B	120.9 (2)
N1A-C14A-C12A	114.3 (2)	N1B - C14B - C12B	115.8 (2)
C2A - O1A - C1A	118.0 (3)	C2B-O1B-C1B	117.2 (2)
C7A - C12A - C14A - O2A	71.6(2)	C7B - C12B - C14B - O2B	75.6 (2)
C7A - C12A - C14A - N1A - 106.6 (2)		C7B-C12B-C14B-N1B	-103.6(2)
C12A - C14A - N1A - C15A 171.3 (2)		C12B - C14B - N1B - C15B	178.7 (2)
C17A - C15A - N1A - C14A	4 114.6 (3)	C16 <i>B</i> – C15 <i>B</i> – N1 <i>B</i> – C14 <i>B</i>	-125.4(3)
C16A - C15A - N1A - C14A - 120.5 (3)		C17 <i>B</i> -C15 <i>B</i> -N1 <i>B</i> -C14 <i>B</i>	111.4 (3
C3A-C2A-O1A-C1A	5.5 (5)	C3B-C2B-O1B-C1B	-4.0(4)

 Table 2

 Hydrogen-bonding geometry (\mathring{A} , °).

D $ H$ $\cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
$ \begin{array}{c} N1A - H25A \cdots O2B \\ N1B - H25B \cdots O2A^{i} \end{array} $	0.97 (3)	1.96 (3)	2.911 (3)	166 (2)
	0.98 (3)	2.09 (3)	3.051 (3)	166 (2)

Symmetry code: (i) 1 + x, y, 1 + z.

H atoms were constrained (included as riding atoms) except for HN1A and HN1B which were refined, with isotropic displacement parameters fixed at $1.2U_{\rm eq}$ of the parent atom ($1.5U_{\rm eq}$ for methyl atoms).

Data collection: MAR software; cell refinement: *marHKL* (Klein & Bartels, 2000); data reduction: *marHKL* (Klein & Bartels, 2000); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996); software used to prepare material for publication: *SHELXL*97.

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